

Protein-based microporous carbon nanoplates for supercapacitors

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Despite their high power density, the energy density of supercapacitors is an order of magnitude lower than that of conventional batteries. Advanced carbon-based nanomaterials can improve the energy density of supercapacitors. Supercapacitors can be classified into two categories with respect to their energy storage mechanisms. One is the electric double-layer capacitor, where the capacitance comes from the pure electrostatic charge accumulated at the electrode/electrolyte interface. It requires electrode materials with high surface area and pores adapted to the size of the ions in the electrolyte.^{1,2} The other is the pseudocapacitor, in which fast and reversible faradic processes take place due to electro-active species.³ The incorporation of heteroatoms can considerably improve the specific capacitance of carbon nanomaterials because the heteroatoms induce pseudocapacitive behavior and improve the polar properties of the carbon nanomaterials.³⁻⁵

In this study, novel protein-based microporous carbon nanoplates containing numerous heteroatoms were fabricated from regenerated silk fibroin by the carbonization and activation of KOH. Each carbon nanoplate had a large surface area (2553.3 m²/g), and contained the electro-active heteroatoms nitrogen (5.1 wt%) and oxygen (10.7 wt%). The nanoplates' electrical conductivity at 300 K, 1.15×10^4 S/m, is comparable to that of highly doped silicon. The carbon nanoplates exhibited excellent electrochemical performance, displaying a specific capacitance of 264 F/g in aqueous electrolytes, a specific energy of 133 Wh/kg, a specific power of 217 kW/kg, and a stable cycle life of over 10000 cycles.

References

- [1] P. Simon, Y. Gogotsi, *Nat. Mater.*, 7, (2008) 845.
- [2] L. L. Zhang, X. S. Zhao, *Chem. Soc. Rev.*, 38 (2009) 2520.
- [3] Y. S. Yun, JY. Shim, YS. Tak, H.-J. Jin, *RSC adv.*, 2, (2012) 4353.
- [4] Y. S. Yun, CB. Im, H. H. Park, I. H. Hwang, YS. Tak, H.-J. Jin, *J. Power Sources*, 234, (2013) 285.
- [5] Y. S. Yun, S. Y. Cho, J. Shim, B. H. Kim, S.-J. Chang, S. J. Baek, Y. S. Huh, Y. Tak, Y. W. Park, S. Park, H.-J. Jin, *Adv. Mater.*, 25 (2013) 1993.

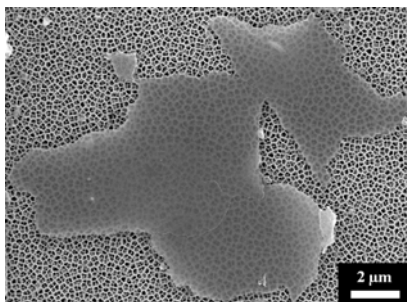


Figure 1. FE-SEM image of the carbon nanoplates on alumina template membranes.